

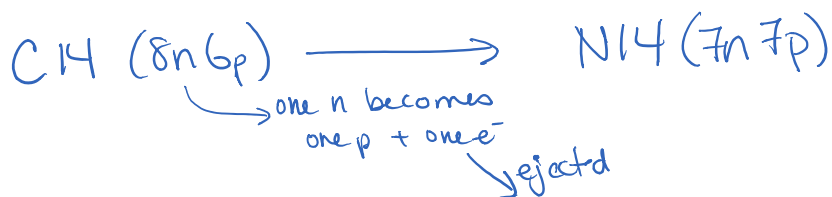
pg 156-158 - Early ideas of age of \oplus

radioactive isotopes - unstable nucleus (parent) so, in time, it decays changing into a different nucleus that is stable (daughter)

[produces heat - believed to provide the heat that has kept \oplus molten inside providing energy for plate tectonics]

radiation of decay - alpha particles (α He nuclei = $2p+2n$)
- beta particles (electrons, e^- , β)
- gamma rays (electromagnetic radiation, strongest energy)

ex carbon 14 (parent) decays by ejecting beta particles (e^-) and forms Nitrogen 14 (daughter)



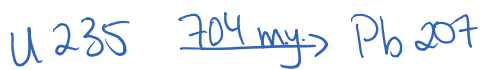
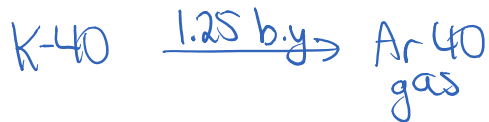
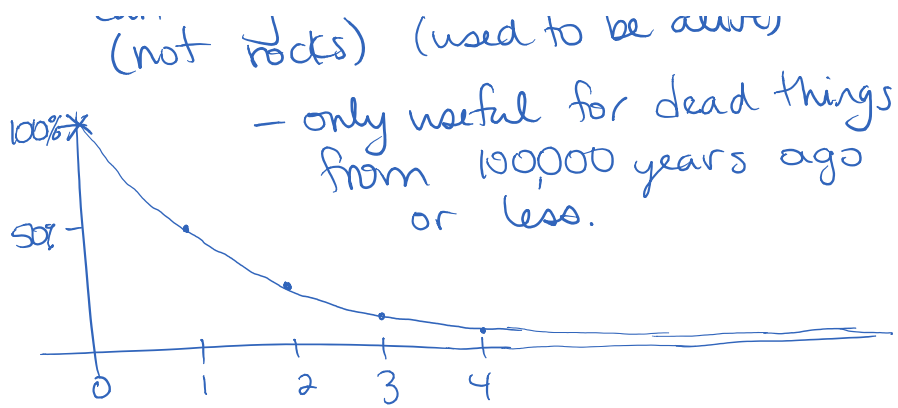
radioactive decay obeys the laws of probability:

half-life - length of time required for half of a given initial # of parent atoms to decay into the daughter isotope

half-life is constant for each element

* C14 h.l. = 5730 year

↑ can only be used for dead things



ex Dead tree sample has 16 Carbon-14 atoms and 48 Nitrogen-14 atoms. How long has tree been dead? h.l. = 5730 yrs

16 + 48 = 64 atoms of C-14 when tree died



$$\text{Age} = \left(\frac{\# \text{ of}}{\text{h.l.}} \right) \left(\frac{\text{length}}{\text{of h.l.}} \right)$$

$$= (2)(5730 \text{ y})$$

$$= 11,460 \text{ years since tree died}$$

Possible Problems

- may have been some daughter products to start with (make rock look older)
- may not be a closed system - for ex. K-Ar, argon is a gas that can escape making rock look younger

Choice of Isotope

- parent was/is abundant in sample
- h.l. appropriate to age being dated
- what are you dating?

dead thing - use C^{14} up to 100,000 y old
rock - any method, except C^{14}

meta. rock - K-Ar since Ar escapes in
meta processes, resetting the K-Ar clock

Absolute Age - numerical age from radiometric
dating, counting tree ring, counting
varves (layers at bottom of glacial lake)

We use a combination of relative and
absolute dating methods.